PREPARATION OF N-ACETYLNORMURAMYL-α-AMINOBUTYRYL--D-ISOGLUTAMINYL-LYSYL-LYSYL-LYSINE*

Viktor Krchňák, Jan Ježek and Milan Zaoral

Institute of Organic Chemistry and Biochemistry, Czechoslovak Academy of Sciences, 166 10 Prague 6

Received July 12th, 1982

The title compound (1) has been prepared by solid-phase synthesis. In aqueous solution the compound 1 forms a mixture of α - and β -anomers.

N-Acetylnormuramyl-\alpha-aminobutyryl-D-isoglutamine^{1,**} inhibits significantly the growth of cholanthrene-induced tumors. A change in acidobasic properties may influence its behaviour in the organism (transport, elimination, metabolism, persistence, accumulation in various tissues, etc.) and thus also its cancerostatic properties. In order to check this assumption, we synthesized N-acetylnormuramyl- α -aminobutyryl-D-isoglutaminyl-lysyl-lysyl-lysine (I). The compound was prepared by solid--phase synthesis^{2,3}. Merrifield resin, esterified⁴ with N^c-p-toluenesulfonyllysine, was condensed successively with two lysine moieties, isoglutamine moiety, α -aminobutyric acid mojety and finally with 1-α-O-benzyl-4.6-O-benzylidene-N-acetylnormuramic acid. The α -amino functions were protected with tert-butyloxycarbonyl group and 1.5 equivalent of 1-hydroxybenzotriazole⁵ was used in the coupling reaction. The condensation of protected N-acetylnormuramic acid⁶ was carried out for 3 h without repeating. The protected glycopeptide was removed from the resin by ammonolysis⁷ and the protecting groups were cleaved off by sodium in liquid ammonia⁸. The free glycopeptide was desalted on a column of Amberlite IRC-50 and purified by high performance liquid chromatography. The product was eluted from the column in two fractions, corresponding to the α - and β -anomers. Equilibration of both fractions in water for 24 h afforded identical mixtures of both anomeric forms. In a delayed type hypersensitivity test, using the 24 M type protein as antigene, the immunoadjuvant activity of compound I was comparable with that of N-acetylnormuramyl-a-aminobutyryl-D-isoglutamine.

Part IV in the series Synthetic Glycopeptides; Part III: This Journal 47, 2989 (1982).

^{**} Unless stated otherwise, the optically active amino acids are of the L-configuration and the sugar component of the D-configuration. Symbols usual in the amino acid and sugar chemistry are employed.

EXPERIMENTAL

Melting points were determined on a Kofler block and are uncorrected. Cptical rotation was measured on a Perkin-Elmer 141 polarimeter. Purity of the products was checked by thin-layer chromatography (Kavalier, Votice, Czechoslovakia; detection by chlorination method^{9,10}) and by paper electrophoresis on a Whatman No 3 MM paper at 700 V (about 50 V/cm) in aqueous acetic acid (pH 2.5); detection with ninhydrin. Samples for the amino acid analysis were dried for 8 h at 10 Pa and 100°C over P₂O₅. The buffered hydrolyzates (pH 2.2) were analyzed on an amino acid analyser Beckman-Spinco, model 120B. The colour value for normuramic acid was 3.06; the values were corrected for 25% decomposition during the hydrolysis¹¹. Preparative HPLC was performed on a 25 × 2 cm column of Separon R with methanol-water-trifluoroacetic acid (5:95:0-2) as the mobile phase, the compounds being detected at 210 mm using a UVM 4 spectrophotometer (Developmental Workshops of Czechoslovak Academy of Sciences, Prague).

$$\label{eq:action} \begin{split} & I-\alpha-O-Benzyl-4, 6-O-benzylidene-N-acetylnormuramyl-\alpha-aminobutyryl-D-isoglutaminyl-N^c-p-toluenesulfonyllysyl-N^c-p-toluenesulfonylly$$

The synthesis was carried out on a chloromethylated polystyrene resin, cross-linked by 2% of divinylbenzene (Calbiochem, Los Angeles, USA), containing 0.96 mmol Cl/g. After esterification⁴ with Boc-Lys(Tos) the resin contained 0.50 mmol Lys/g. The synthesis was performed with 2 g of the resin (1 mmol of Lys) on a manually controlled semiautomatic synthetizer of our own construction. The α -amino groups were protected with Boc group, N^{ε}-amino group in Lys with p-toluenesulfonyl group. $1-\alpha$ -O-Benzyl-4,6-O-benzylidene-N-acetylnormuramic acid⁶ was used as protected sugar component. The couplings were carried out according to the scheme described preciously³ (programme No 6). The protected normuramic acid was coupled for 3 h without repetition; in all the couplings 1.5 equivalent of 1-hydroxybenzotriazole hydrate was employed. Ammonolysis⁷ afforded 1.2 g (79%) of the product, m.p. 201-204°C which upon two crystallizations melted at 207-209°C. The product was chromatographically homogeneous in l-butanol-acetic acid-water (4:1:1) and 1-butanol-acetic acid-pyridine-water (15:3:12:10), $[\alpha]_{D}^{20} + 25\cdot4^{\circ}$ (c 0.26, dimethylformamide). For $C_{72}H_{97}N_{11}O_{19}S_3H_2O$ (1 535) calculated: 56.33% C, 6.50% H, 10.04% N; found: 56.72% C, 6.36% H, 9.97% N. Amino acid composition: NorMur 0.95, Abu 1.02, Glu 1.02, Lys 2.40 (8 h hydrolysis); Abu 1.03, Glu 1.01, Lys 2.96 (20 h hydrolysis).

N-Acetylnormuramyl- α -aminobutyryl-D-isoglutaminyl-lysyl-lysyl-lysine Amide (I)

The protected glycopentapeptide *II*, m.p. $201-204^{\circ}$ C (200 mg; 0·13 mmol) was reduced with sodium in liquid ammonia and desalted on a column of Amberlite IRC 50 according to ref.¹¹, affording 174 mg (76%) of lyophilizate which on electrophoresis exhibited three minor impurities (about 10%). Purification by preparative HPCL yielded 87 mg (38%) of product, $[\alpha]_D^{20} + 21\cdot2^{\circ}$ (after 24 h 19·3°) (c 0·2; water), nitrogen content 15·12% (the lyophilizate contained 86% of compound *I*). Amino acid composition: NorMur 0·86, Abu 1·10, Glu 1·06, Lys 2·97 (8 h hydrolysis).

REFERENCES

- Laughlin C. A., Schwartzman S. M., Horner B. L., Jones G. H., Moffatt J. G., Nestor J. J. jr, Tegg D.: Science 208, 415 (1980).
- 2. Merrifield R. B.: J. Amer. Chem. Soc. 85, 2149 (1963).
- 3. Krchňák V., Zaoral M.: This Journal 44, 1173 (1979).

- 4. Horiki K., Igano K., Inouye K.: Chem. Lett. 1978, 165.
- 5. König W., Geiger R.: Chem. Ber. 103, 788 (1970).
- 6. Chaturvedi N. C., Khosla M. C., Anand N.: J. Med. Chem. 9, 911 (1966).
- 7. Bodanszky M., Sheehan J. T.: Chem. Ind. (London) 1964, 1423.
- 8. Zaoral M., Ježek J., Straka R., Mašek K.: This Journal 43, 1797 (1978).
- 9. Reindel F., Hoppe W.: Chem. Ber. 87, 1103 (1954).
- Brenner M., Zimmermann J. P., Wehrmüller J., Quit P., Hartmann A., Schneider W., Beglinger U.: Helv. Chim. Acta 40, 1497 (1957).
- 11. Zaoral M., Ježek J., Krchňák V., Straka R.: This Journal 45, 1424 (1980).

Translated by M. Tichy.